## **Year 12**

## **Subject: Chemistry**

Time period	Teacher A Physical Chemistry	Teacher B Organic/ Inorganic	Required practical/Notes
Autumn 1 2023	<b>3.1.2 Amount of substance</b> (Kerboodle chapter 2) Relative atomic and molecular masses, Moles in solutions, The ideal gas equation, Empirical and molecular formul, Atom	<ul> <li>3.3.1 Introduction to organic chemistry (Kerboodle chapter 11)- Nomenclature, Isomerism</li> <li>3.3.2 Alkanes (Kreboodle chapter 12) -</li> </ul>	<u>RP1</u> Make up a volumetric solution and carry out a simple acid-base titration
	economy, % yield, concentrations	Fractional distillation, cracking, combustion, formation of halogenoalkanes	Assessment
Autumn 2 2023	<ul> <li>3.1.2 Amount of substance - (Kerboodle chapter 2), to be completed</li> <li>3.1.1 Atomic structure (Kerboodle chapter 1) – mass number, atomic number, isotopes,</li> </ul>	<ul> <li>3.3.3 Halogenoalkanes</li> <li>(Kerboodle chapter 13) - Nucleophilic substitution, elimination reactions</li> <li>3.3.4 Alkenes</li> </ul>	w/c 27/11-Learning judgement + Forecast Grade
	<ul> <li>the arrangement of electrons, the mass spectrometer, Ionisation energy</li> <li><b>3.1.3 Bonding</b> (Kerboodle chapter 3) – Covalent, metallic bonding, electronegativity, forces acting between molecules, the shapes of molecules, bonding and physical properties</li> </ul>	(kerboodle chapter 14) – Reactions of alkenes, addition polymers	Assessment
Spring 1 2024	<b>3.1.3 Bonding</b> (Kerboodle chapter 3) – continue	<b>3.3.5 Alcohols</b> (Kerboodle chapter 15) – Ethanol production, the reactions of alcohols	<u>RP5</u> Distillation of a product from a reaction
	<b>3.1.4 Energetics</b> (Kerboodle chapter 4) – Exothermic and endothermic reactions,		<u>RP2</u> Measurement of an enthalpy change

## HIGHGATE WOOD SCHOOL: CURRICULUM MAP FOR KEY STAGE 5

	enthalpy, Hess's Law, Enthalpy changes of combustion, bond enthalpies, Thermochemical cycles)	<b>3.3.6 Organic analysis</b> (Kerboodle chapter 16) – test -tube reactions, Mass spectrometry, IR spectroscopy	Assessment
Spring 2 2024	<b>3.1.4 Energetics</b> (Kerboodle chapter 4) - continue	<b>3.3.6 Organic analysis</b> (Kerboodle chapter 16) to be completed	<u>RP6</u> Tests for alcohol, aldehyde, alkene and carboxylic acid
	<b>3.1.5 Kinetics</b> (Kerboodle chapter 5) – Collision theory, The Maxwell-Boltzmann distribution, Catalyst	<b>3.2.1 Periodicity</b> (Kerboodle chapter 8) – The Periodic table, trends in the properties of elements Period 3, Ionisation energies	<u>RP3</u> Investigation of how the rate of a reaction changes with temperature
		<b>3.2.2 Group 2 The Alkaline Earth Metals</b> (Kerboodle chapter 2) – The physical and chemical properties of Group 2	Assessment w/c 11/03 – Learning judgement + Forecast Grade
Summer 1 2024	<ul> <li>3.1.6 Equilibria (Kerboodle chapter 6) – Changing the condition of an equilibrium reaction, The equilibrium constant, Kc</li> <li>3.1.7 Oxidation, reduction, redox (Kerboodle chapter 7) – Oxidations states, redox equations</li> </ul>	<b>3.2.3 Group 7(17), The Halogens</b> (Kerboodle chapter 10)- The chemical reactions of the halogens, reactions of halide ions, use of chlorine	RP4 Carry out simple test-tube reactions to identify cations and anions in aqueous solution w/c 15/04-22/04/24- MOCK exams w/c 13/05 – Learning judgement+ Mock + Forecast Grade
Summer 2 2024	<ul> <li>3.1.7 Oxidation, reduction, redox (Kerboodle chapter 7) to be completed</li> <li>Start A2</li> <li>3.1.8. Thermodynamics (Kerboodle chapter 17)</li> </ul>	Start A2 3.3.7 Optical isomerism (Kerboodle chapter 25)	Assessment